

Original Article



The Antimicrobial and Antibiofilm Potential of *Citrus Aurantium* and *Artemisia Annua* Essential Oils Nanoemulsions

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ABSTRACT

Antimicrobial resistance has posed considerable health and economic burdens globally (approximately five million deaths annually), particularly in developing countries. The estimated annual treatment costs in the United States include US\$4.6 billion. Vast antibiotic resistance among Gram-negative and Gram-positive bacterial species has spread from healthcare to the environment, community, and animals. These conditions have limited and in some cases, failed infection eradication options and facilitated the distribution of drug-resistant organisms. The spread of drug-resistant bacterial infections is a major human health concern, hence, seeking novel antibacterial agents is crucial. This study used nanoemulsions of *Citrus aurantium* and *Artemisia annua* essential oils (EOs) as natural antibacterial agents. Gas chromatography mass spectrometry (GC-MS) analysis showed that limonene (31.4%) and artemisia ketone (26.2%) were major components, respectively. After that, their nanoemulsion dosage forms with mean droplet sizes of 181 ± 7 and 160 ± 5 and zeta potential values 3.1 ± 0.8 and -4.9 ± 0.5 mV were prepared. Meanwhile, successful loading of the EOs in nanoemulsion was confirmed by Attenuated Total Reflection-Fourier Transform Infrared (ATR-FTIR) analysis. *A. annua* nanoemulsion with 40% antioxidant effect was significantly more potent than *C. aurantium* nanoemulsion. Meanwhile, nanoemulsions' antibacterial and antibiofilm activity against clinical and standard strains, *Escherichia coli*, *Staphylococcus aureus*, *Pseudomonas aeruginosa*, and *Klebsiella pneumoniae*, were investigated. The best efficiency was related to the effect of *C. aurantium* nanoemulsion against *S. aureus*; minimum inhibitory (MIC) and minimum bactericidal concentrations (MBC) were 500 and > 2000 $\mu\text{g/mL}$. In addition, no biofilm was formed after treatment with both nanoemulsions. Therefore, *C. aurantium* and *A. annua* EO nanoemulsions may act as natural antioxidant and antibacterial agents in complementary medicine.

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1. Introduction

Vast antibiotic resistance among Gram-negative and Gram-positive bacteria has spread from healthcare to the environment, community, and animals. These conditions have limited, and in some cases, failed the infection eradication options, facilitating the distribution of drug-resistant organisms (1, 2) such as those non-susceptible to last-resort antibiotics of carbapenems and glycopeptides. As alternatives to chemotherapies, natural resources such as essential oils (EOs) have been suitable alternatives (3-5). In addition, nanoformulation of EOs by optimization in particle size, increasing solubility, and improving bioavailability and stability causes medicines to be more permeable into cells, such as bacteria or cancer cells (6-10).

The *citrus* genus includes several species of citrons within the family *Rutaceae*, used as herbal medicines, fruits, juice, and additives. *Citrus* fruits contain vitamins C and B, minerals, nutrients, and bioactive compounds such as phenolic compounds, volatile oils, and terpenoids. *Citrus aurantium* (*C. aurantium*) L. cultivar has exhibited anti-inflammatory, hypoglycemic, antimicrobial, anticancer, pain-relief, and organ-protective effects (11). Although the species has bioactive compounds and biological activities, its pharmacological effects, traditional usage, and exact bioactive compounds have not been uncovered (12). On the other hand, *C. aurantium* L. dietary supplementation has not exerted side effects (13). Various extracts of *C. aurantium* L. leaves, including aqueous, alcoholic, or chloroform portions, have demonstrated antibacterial effects against various agents (11, 13).

In addition, the antimicrobial effects of its EOs have unraveled potential activities against *Agrobacterium tumefaciens*, *Dickeya solani*, and *Erwinia amylovora* (14, 15). *Artemisia annua* grows globally in Europe, Asia and North America with preferred arid and semi-arid climates and provides considerable health benefits such as anticancer and antimicrobial properties. Major bioactive compounds include artemisia ketone, 1,8-cineole, germacrene D, and camphor (16). This study aimed to investigate the antibacterial and antibiofilm effects of nanoemulsions of *A. annua* L. and *C. aurantium*.

2. Materials and Methods

2.1. Materials

C. aurantium and *A. annua* EOs were supplied from Iranian companies, Tabib Daru Company and Pharmaceutical Company Essential Oil Dr. Soleimani. Bacterial standard strains included *Escherichia coli* ATCC25922, *Staphylococcus aureus* ATCC25923, *Pseudomonas aeruginosa* ATCC 27853, and *Klebsiella pneumonia* ATCC13883 were cultured from bacterial stock preserved at -20 °C in the microbiology laboratory of Fasa University of Medical Sciences, Iran. Moreover, clinical isolates were also obtained from patients. Mueller Hinton broth, trypticase soy broth, crystal violet, tween 20, and tween 80 were purchased from Merck Chemicals, Germany. The standard analytical solution of Alkanes (C9-C24) was supplied from Sigma-Aldrich (USA).

2.2. Chemical Composition of Eos

Gas chromatography mass spectrometry (GC-MS) analysis was used for the chemical compositions of the EOs. For this purpose, the gas chromatographic device (Agilent 6890, USA) with HP-5MS silica fused columns coupled to a network mass selective detector (Agilent 5973, USA) was used. Major constituents of EOs were identified by comparing their retention indices to homologous C9-C24 n-alkanes, as previously described (17, 18).

2.3. Preparation and Characterizations of Nanoemulsions

A. annua EO (0.4% v/v) was mixed with tween 80 (0.5% v/v), and *C. aurantium* (0.4% v/v) was mixed with tween 20 (0.5% v/v), separately, for 3 minutes at room temperature at 2000 rpm. Distilled water was then added dropwise to reach the final volume of 5 mL, and the mixture was stirred for 40 minutes at room temperature at 2000 rpm. Droplet size and droplet size distribution (SPAN) were measured using a DLS (Dynamic Light Scattering, DLS9900, K-ONE, Korea) device. SPAN was calculated using the relationship $d_{90}-d_{10}/d_{50}$; in this equation, d is the diameter, 90, 10, and 50 percent of particles with a size smaller than the values mentioned.

The droplet size below 200 nm and SPAN below 1 nm were necessary conditions to confirm the appropriate size characteristics. TEM (Transitional Electron Microscopy, Philips, TEM, EM 208s, Netherland) was used to confirm the droplet size and determine their morphology. ATR-FTIR (Attenuated Total Reflection-Fourier Transform Infrared) analysis is used to evaluate the successful loading of the EOs in the nanoemulsion. Spectra of the EOs, nanoemulsion (-oil), and nanoemulsion were

recorded in 400 - 4000 cm^{-1} using a spectrometer (Tensor II model, Bruker Co, Germany).

Furthermore, the stability of nanoemulsions was investigated. Nanoemulsions were centrifuged at -4 , $+4$, and $+25^\circ\text{C}$ (14,000 g, 30 min) to investigate stability against precipitation. Additionally, nanoemulsions were stored at $+45^\circ\text{C}$ and room temperature for six consecutive intervals of 48 hours for thermal stability analysis. Moreover, nanoemulsions were placed at -20°C and room temperature for six consecutive 48-hour intervals for cryogenic stability. In addition, nanoemulsions were placed at 4°C and room temperature for six months for long-term stability analysis. After each test, the nanoemulsion was visually checked for sedimentation, creaming, or phase separation.

2.4. Investigation of Antioxidant Properties of EOs and Nanoemulsions

DPPH (2,2-diphenyl-1-picrylhydrazyl) assay was used to measure antioxidant properties. First, serial dilutions of the nanoemulsions (62.5-2000 $\mu\text{g}/\text{mL}$) were prepared in ethanol. Next, 50 μL /well of each prepared dilution and 0.2 mM DPPH solution was added to a 96-well plate, and was incubated for 30 minutes in the dark at room temperature. Finally, the wells' OD (optical density) was measured at 517 nm using a plate reader (Synergy HTX Multi-Model Reader, USA). Antioxidant activity was calculated using $\text{OD test}/\text{OD control} \times 100$.

2.5. Minimum Inhibitory Concentration

Micro-dilution test was implemented. The range of concentrations of nanoemulsions (250, 500, 1000, 1500 and 2000 $\mu\text{g}/\text{mL}$) was prepared by PBS containing 0.5% DMSO as solvent. Antibacterial effects of EOs and nanoemulsion were investigated using 96-well broth micro-dilution, as previously described(19). Briefly, 40 μL of each was inoculated into wells of 96 well plates containing 50 μL of Mueller Hinton broth. Afterward, 10 μL /well of each bacterial suspension (0.5 McFarland standard turbidity, 1.5×10^8 CFU/mL) was added to each well. The plates were incubated for 24 hours at 37°C , and then the OD of the wells was read at 630 nm. Bacterial growth was calculated using $\text{OD sample}/\text{OD control} \times 100$. The minimum inhibitory concentration (MIC) of *A. annua* and *C. aurantium* single and nanoemulsion forms against bacterial strains were determined using concentrations ranging from 250-2000 $\mu\text{g}/\text{mL}$. Moreover, the bacterial suspension equal to 0.5 McFarland standard was prepared. The test was performed the same as the broth micro-dilution method.

2.6. Biofilm Formation

Anti-biofilm effects of EOs and nanoemulsions were assessed against clinical isolates. The biofilm formation with and without exposure to the nanoemulsions was performed into 96-well plates using a microtitre tissue plate assay. Briefly, an overnight culture of bacterial strains was obtained into the trypticase soy broth (TSB) medium containing 1% glucose and diluted 1:100. For each bacterial suspension, 20 μL was taken and inoculated into wells containing 180 μL of the TSB medium in triplicate and incubated for 5 hours for exposed (2000 $\mu\text{g}/\text{mL}$ of each nanoemulsion) group and 24h for the unexposed group.

The unexposed group medium was exchanged with each nanoemulsion and incubated for 24 hours. Next, the wells were washed using double distilled water and fixed using methanol. Then, 0.1% crystal violet or safranin was added for 15 min. After washing, the absolute ethanol (200 μL) was added to solubilize bacterial contents and read using the ELISA reader at 490 nm. The biofilm formation levels (strong, moderate, weak, or non-adherence) were calculated using Table 1. This study compared biofilm formation levels among groups, including the control (unexposed) and *A. annua* and *C. aurantium* EOs nanoemulsion-treated groups.

2.7. Data analysis

All experiments were done in triplicates. The data were analyzed using the SPSS software, from which Chi-Square and analysis of variance (ANOVA) tests were applied to determine differences at a *p*-value cut-off of 0.05.

3. Results

The main components of the *C. aurantium* were limonene (31.4%), sabinene (15.6%), γ -terpinene (6.0%), linalool (5.6%) and cis-nerolidol (5.1%) (Table 2). Similarly, the main components of the *A. annua* were artemisia ketone (26.2%), camphor (19.2%), 1,8-cineole (12.3%), *trans*-caryophyllene (4.5%), and camphene (4.4%) (Table 2).

3.1. TEM analysis of *C. Aurantium* and *A. Annua* EO Nanoemulsions

The mean droplet diameter and zeta potential of *C. aurantium* nanoemulsion were 181 ± 7 nm and 3.1 ± 0.8 mV, respectively (Figures 1 A & B). The *A. annua* nanoemulsions exhibited a droplet diameter of 160 ± 5 nm and zeta potential values of -4.9 ± 0.5 mV (Figures 2 A & B).

Table1. The calculation of biofilm formation levels

Biofilm formation ability	Calculation of cut-off level	OD calculated results	Reference
Strong	$OD > OD_c \times 4$	$0.33296 > OD$	(20, 21)
Moderate	$OD_c \times 2 \leq OD < OD_c \times 4$	$0.16648 \leq OD < 0.33296$	
Weak	$OD_c \leq OD < 2 \times OD_c$	$0.083324 \leq OD < 0.16648$	
No binding	$OD \leq OD_c$	$0.08324 \leq OD$	

OD: optical density, OD_c: mean OD of control wells.

Table 2. Identified compounds in the EOs using GC-MS analysis.

Retention Time (min)	Compound	<i>C. aurantium</i>		<i>A. annua</i>		Retention Index
		Area	%	Area	%	
9.46	α -pinene	58484827	1.7	122296938	4.1	932
10.06	camphene	--	--	132788539	4.4	954
11.14	sabinene	542668432	15.6	44039171	1.5	975
11.24	β -pinene	43301938	1.2	29556997	1.0	979
11.96	β -myrcene	108784770	3.1	29379228	1.0	988
12.50	Yomogi alcohol	--	--	41103264	1.4	999
13.08	α -terpinene	59177141	1.7	--	--	1014
13.81	1,8-cineole	--	--	368453758	12.3	1026
13.89	limonene	1088445097	31.4	--	--	1029
14.67	<i>cis</i> -ocimene	162728160	4.7	--	--	1037
15.13	γ -terpinene	207125216	6.0	--	--	1054
15.48	artemisia ketone	--	--	784989266	26.2	1062
16.31	artemisia alcohol	--	--	33487881	1.1	1083
17.16	linalool	192637034	5.6	--	--	1095
19.22	camphor	--	--	576552736	19.2	1146
20.12	borneol	--	--	28483988	1.0	1169
20.63	4-terpineol	66692764	1.9	33586786	1.1	1177
23.55	cuminic aldehyde	128003231	3.7	--	--	1239
29.38	α -copaene	--	--	47639328	1.6	1376
31.26	<i>trans</i> -caryophyllene	--	--	136219018	4.5	1419
33.78	germacrene D	--	--	84811438	2.8	1481
33.99	β -selinene	--	--	90133377	3.0	1490
37.11	<i>cis</i> -nerolidol	178234586	5.1	--	--	1532
37.76	caryophyllene oxide	--	--	35663606	1.2	1583
42.89	<i>cis</i> -farnesol	34776356	1.0	--	--	1698

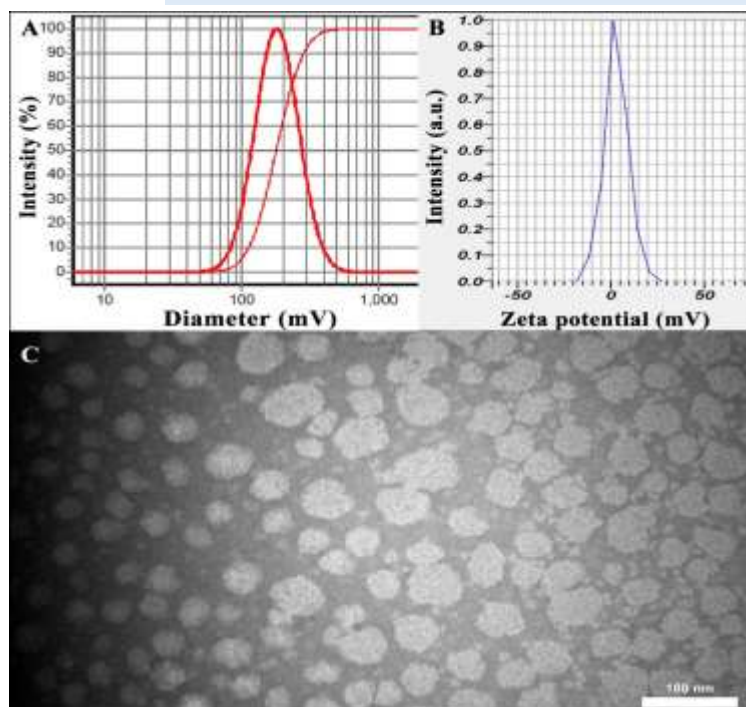


Figure 1. Characterization of *C. aurantium* EO nanoemulsion, A: DLS profile, B: zeta potential profile, and C: TEM image.

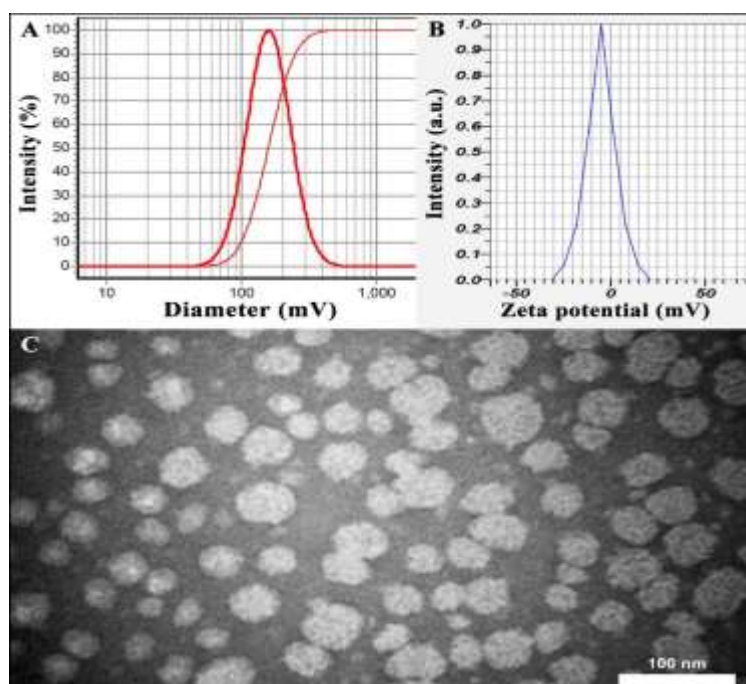


Figure 2. Characterization of *A. annua* EO nanoemulsion, A: DLS profile, B: zeta potential profile, and C: TEM image.

The TEM analysis (Figures 1C and 2C) revealed that both nanoemulsions were spherical in shape, with sizes < 100 nm. Furthermore, the stability of nanoemulsions was investigated.

They were centrifuged at -4 , $+4$, and $+25^{\circ}\text{C}$ (14,000 g, 30 min); no sedimentation or phase separation was observed. Additionally, nanoemulsions were stored at $+45^{\circ}\text{C}$ and room temperature for six consecutive intervals of 48 hours for thermal stability analysis; no sedimentation or bi-phasic condition was observed.

Moreover, nanoemulsions were placed at -20°C and room temperature for six consecutive 48-hour intervals for cryogenic stability; no sedimentation or bi-phasic condition was observed. In addition, nanoemulsions were placed at 4 and room temperature for six months for long-term stability analysis; no sedimentation or phase separation was observed.

3.2. ATR-FTIR Analysis of *C. Aurantium* And *A. Annua* Nanoemulsions

ATR-FTIR analysis confirmed EO loading in nanoemulsion (Figure 3). The spectra of *C. aurantium*

EO, displayed in Figure 3A, broadband at 3469 cm^{-1} can be attributed to stretching vibration of the hydroxyl group due to hydrogen bonding in alcoholic and phenolic bioactive compounds in EO, spectra at 3076 cm^{-1} can be corresponded to stretching vibration of CH in sp^2 groups and the bands at 2961 , 2923 and, 2872 cm^{-1} can be attributed to stretching vibration of CH in sp^3 groups, the band at 1706 and 1676 cm^{-1} can be attributed to stretching vibration of carbonyl groups.

Bands at 1108 and 1052 cm^{-1} showed stretching vibration of C-O groups. The peak at 989 cm^{-1} is attributed to C-H bending absorption, and the strong peak at 758 cm^{-1} is attributed to benzene rings C-H vibration absorption.

A peak at 689 cm^{-1} is attributed to the vibration absorption of alkenes. The spectrum of nanoemulsion without *C. aurantium* EO displayed in Figure 3B, the broad peak at about 3493 cm^{-1} can be attributed to OH stretching vibration due to hydrogen bonding between water and tween 20. Spectra at 2924 cm^{-1} corresponded to C-H stretching in tween 20. A strong band at 1733 cm^{-1} attributed to C=O stretching, representing the carbonyl group in tween 20.

Characteristic band at around 1462 cm^{-1} can be attributed to CH_2 bending tween 20. Characteristic and sharp peak at 1091 cm^{-1} is assigned to C-O stretching. FTIR of *C. aurantium* EO nanoemulsion spectrum (Figure 3C) showed the broadband at 3518 cm^{-1} attributed to OH stretching vibration due to the strong hydrogen bonding between water, tween 20, and phenolic and alcoholic compounds in EO. Any band at 2969 , 2924 , and 2856 cm^{-1} is related to C-H stretching due to sp^3 hybrid in tween 20 and EO. Strong band at 1728 cm^{-1} showed carbonyl stretching (C=O) tween 20 and EO. The absorption at around 1456 cm^{-1} corresponded to CH_2 bending tween and EO. A sharp and strong peak at about 1093 cm^{-1} can be attributed to C-O stretching.

Spectrum of the *A. annua* EO has been demonstrated in Figure 3D, a broad and characteristic band at about 3520 cm^{-1} , can be attributed to the hydroxyl functional groups in EO, and a band at 3084 cm^{-1} , allocated to the stretching vibration of =C-H groups from olefins in sp^2 hybrid. Peaks at 2963 , 2928 , and 2872 cm^{-1} , related to stretching vibrations of -CH in sp^3 hybrid, the spectra at around 1743 cm^{-1} related to C=O, the absorption around 1620 and 1415 cm^{-1} assigned to C=C, the bands at 1215 and 1167 cm^{-1} are related to (C-O-C) bonds and the band at 876

cm^{-1} can be allocated to angular deformations of CH_2 groups. Spectra of nanoemulsion without *A. annua* EO has been shown in Figure 3E.

A characteristic and broad peak between 3200 to 3600 cm^{-1} corresponds to OH stretching vibration due to the hydrogen bonding between water and tween 80. Spectra at 2964 and 2925 cm^{-1} are related to C-H stretching. Absorption at 1740 cm^{-1} was attributed to carbonyl stretching (C=O) in tween 80 and at about 1456 cm^{-1} can be allocated to CH_2 bending. A strong and characteristic band at 1080 cm^{-1} corresponded to C-O stretching.

ATR-FTIR of nanoemulsion containing *A. annua* EO has been displayed in Figure 3F. A peak at about 3422 cm^{-1} is attributed to OH stretching vibration due to hydrogen bonding between EO, tween 80, and water and at 2924 cm^{-1} corresponds to C-H stretching of EO and tween 80. Absorption at 1733 cm^{-1} attributed to C=O stretching representing the carbonyl group in EO and tween. A characteristic peak at about 1462 cm^{-1} outlined CH_2 bending in EO and tween 80. A strong and sharp peak at 1091 cm^{-1} was assigned to C-O stretching. The presence of other bands in EOs and blanks confirmed the successful loading of EOs in the prepared nanoemulsion.

The *C. aurantium* and *A. annua* nanoemulsions were evaluated for their antioxidant effects by DPPH assay. As shown in Figure 4, the most potent free radical scavenging activity was obtained from *A. annua* EO nanoemulsion, 40 % at $2000\text{ }\mu\text{g/mL}$.

3.3. The Minimum Inhibitory Concentration (MIC) and Minimum Bactericidal Concentration (MBC)

According to Table 3, the *C. aurantium* EO nanoemulsion MIC against *S. aureus*, *E. coli*, *P. aeruginosa*, and *K. pneumonia* included $500\text{ }\mu\text{g/mL}$, $1000\text{ }\mu\text{g/mL}$, $1000\text{ }\mu\text{g/mL}$, and $1000\text{ }\mu\text{g/mL}$, respectively. The MBC values against bacterial strains also included $>2000\text{ }\mu\text{g/mL}$. Moreover, the MIC values of *A. annua* nanoemulsion EO against *S. aureus*, *E. coli*, *P. aeruginosa*, and *K. pneumonia* included $1000\text{ }\mu\text{g/mL}$, $2000\text{ }\mu\text{g/mL}$, $2000\text{ }\mu\text{g/mL}$, and $2000\text{ }\mu\text{g/mL}$, respectively. The MBC values included $>2000\text{ }\mu\text{g/mL}$ for all the tested bacterial strains.

3.4. Antibacterial Effects

The bacterial growth inhibitory effect of *C. aurantium* nanoemulsion EO was concentration-dependent. The highest bactericidal effect was observed against *S. aureus* at $2000\text{ }\mu\text{g/mL}$, in which 56% of growth was inhibited (Figure 5).

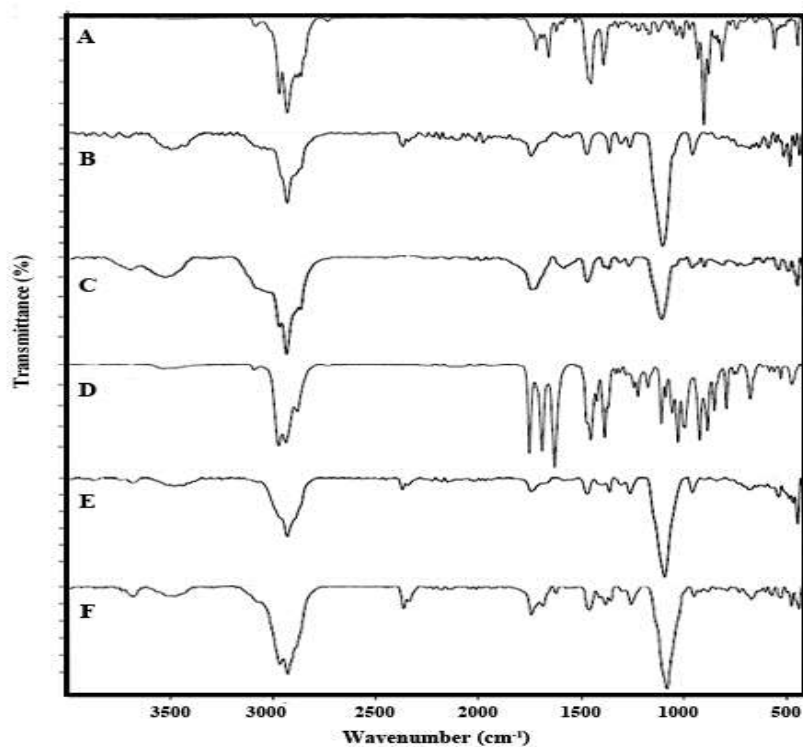


Figure 3. ATR-FTIR spectra of A: *C. aurantium* EO, B: nanoemulsion without *C. aurantium* EO, C: nanoemulsion containing *C. aurantium* EO, D: *A. annua* EO, E: nanoemulsion without *A. annua* EO, F: nanoemulsion containing *A. annua* EO.

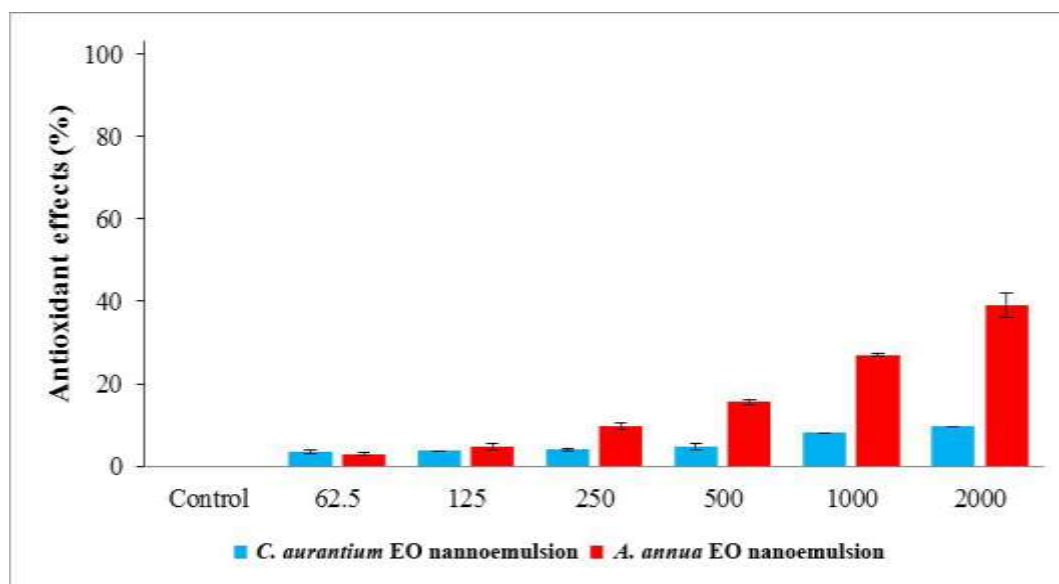


Figure 4. The antioxidant effects of samples.

Table 3. The MIC and MBC levels (µg/mL) samples.

Samples	<i>S. aureus</i> (MIC, MBC)	<i>E. coli</i> (MIC, MBC)	<i>P. aeruginosa</i> (MIC, MBC)	<i>K. pneumonia</i> (MIC, MBC)
<i>C. aurantium</i> EO nanoemulsion	500, >2000	1000, >2000	1000, >2000	1000, >2000
<i>A. annua</i> EO nanoemulsion	1000, >2000	2000, >2000	2000, >2000	2000, >2000

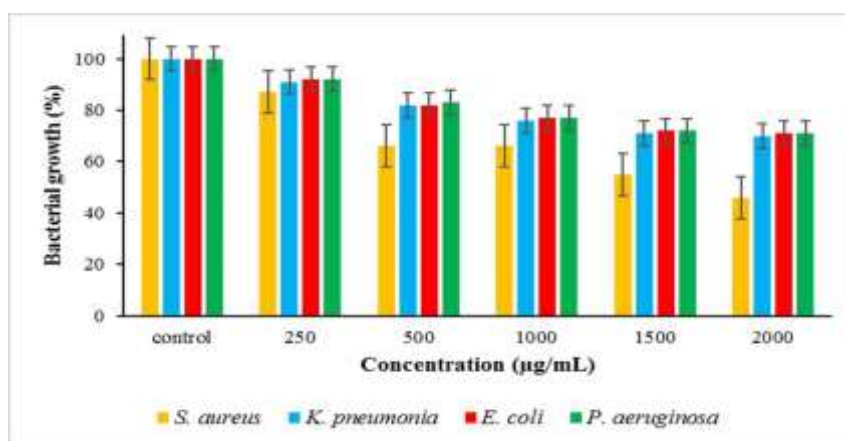


Figure 5. The bacterial growth in exposure to *C. aurantium* nanoemulsion EO.

The growth inhibitory effect of *A. annua* nanoemulsion EO was mostly against *S. aureus* at 2000 µg/mL, in which ~20 % of bacterial growth was inhibited (Figure 6).

3.5. Anti-biofilm effects

As shown in Figure 7, biofilms in the control group all bacteria were formed (OD > 0.083). However, after treatment with both nanoemulsions, no biofilm (OD < 0.083) was formed by all examined bacteria, i.e., *S. aureus*, *K. pneumonia*, *E. coli*, and *P. aeruginosa*.

4. Discussion

In our study, nanoemulsions of *C. aurantium* and *A. annua* EOs were prepared using spontaneous emulsification. Their antimicrobial and antibiofilm against selected bacterial pathogens were subsequently investigated. While previous studies have investigated antimicrobial and insecticidal properties of *C. aurantium* and *A. annua* extracts and EOs against different microorganisms, but our study has focused mostly on their nanoemulsion forms (20).

For example, Kačániová et al examined the biological activity and antibiofilm molecular profile of *C. aurantium* EO, concluding that *C. aurantium* EO exhibited potent antibacterial activity against *Stenotrophomonas maltophilia* and *Bacillus subtilis*, followed by *Penicillium crustosum* (15). In addition, Madhuri et al reported that EOs extracted from the peel of *C. aurantium* can be used against infectious agents such as *K. pneumonia* and *Bacillus cereus* (21).

In vitro studies by Marinas et al. also showed that *A. annua* EO has selective antipathogenic activity on Gram-positive and Gram-negative bacterial strains (22).

Its antibacterial and anti-biofilm activities have been also unveiled (23, 24). In addition, Mariadosset et al. synthesized the selenium nanoparticles using *A. annua* (AaSeNPs) of 109.2 nm in size and the characterized AaSeNPs indicated an antibacterial activity against multidrug-resistant pathogens such as *S. aureus*, *B. subtilis*, *Proteus mirabilis*, and *E. coli* (25). Das et al. also reported stronger antimicrobial activity of Pickering nanoemulsion of the *Artemisia* essential oil (24).

In our study, antimicrobial and antibiofilm effects of nanoemulsions of *C. aurantium* and *A. annua* EOs against selected bacterial pathogens were confirmed through MIC and MBC tests with the highest activity against *S. aureus*. Exposure of pathogens to the nanoemulsions resulted in a significant reduction in the biofilm formation by all examined bacteria, i.e., *S. aureus*, *K. pneumonia*, *E. coli*, and *P. aeruginosa*. This study is the first to investigate the chemical composition of *C. aurantium* and *A. annua* EO in detail, followed by the preparation of their nanoemulsion dosage forms.

The antimicrobial and antibiofilm activities of these nanoemulsions were demonstrated. Both nanoemulsions inhibited the growth of *S. aureus* and notably, no biofilm formations by *S. aureus*, *K. pneumonia*, *E. coli*, and *P. aeruginosa* were observed after treatment with these nanoemulsions.

Moreover, *A. annua* nanoemulsion inferred potent antioxidant properties. Limitations of this study mostly included the low number of bacterial pathogens, the absence of *in silico* assessment of binding of bioactive compounds to bacterial targets and the lack of *in vivo* study.

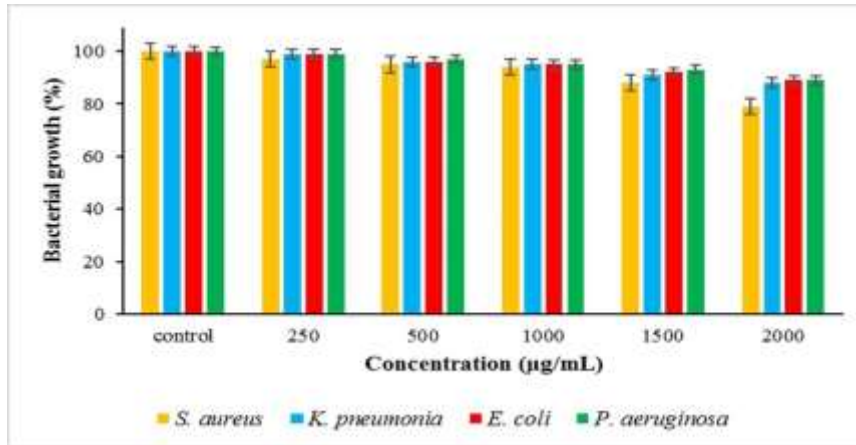


Figure 6. The bacterial growth in exposure to *A. annua* nanoemulsion EO.

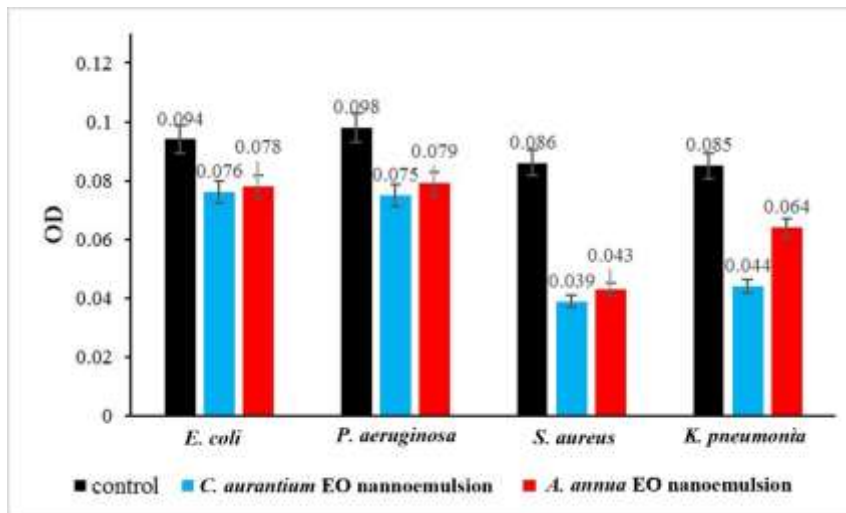


Figure 7. The anti-biofilm effects of (mean OD value) samples.

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Authors' Contribution

Study concept and design: MO, AG, ATR-FTIR.

Acquisition of data: AG, MS.

Analysis and interpretation of data: HA, EZ.

Drafting of the manuscript: MS, MO.

Critical revision of the manuscript for important intellectual content: MAH, BM.

Statistical analysis: DPPH, MO.

Administrative, technical, and material support: MS, AG.

Ethics

This study was supported by Fasa University of Medical Sciences under grant No. 401268 and was ethically approved (IR.FUMS.REC.1401.205).

Conflict of Interest

None.

Funding

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Data Availability

The data supporting the findings of this study are available from the corresponding author upon request.

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